



The Starting Materials

Part I: The Origins of the Elements

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The Starting Materials

Part I: The Origins of the Elements

- Big Bang nucleosynthesis
- Nucleosynthesis in stars
 - Key nucleosynthesis processes
 - Stellar evolution related to nucleosynthetic processes
- Presolar grains (stardust) in meteorites and cosmic dust
 - Stellar sources
 - Probes of specific nuclear reactions
 - Stardust mineralogy – their Galactic history

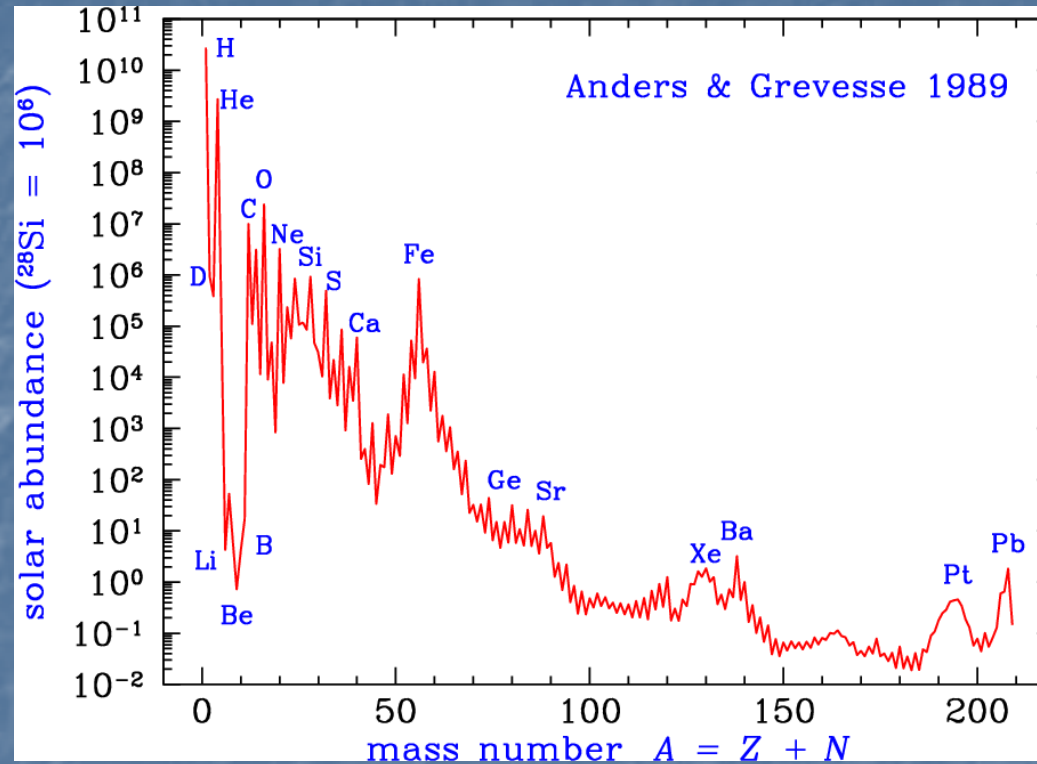
Big Bang Nucleosynthesis

The First Atoms

- Almost all fundamental particles (protons, electrons, neutrons) originated within the first few seconds
- ~1 minute later the Universe cooled to $<10^9$ Kelvin, allowing atomic nuclei to form
- All of the Hydrogen, and most of the Helium and Lithium in the Universe formed within the first minute.
- NO heavy elements formed in the Big Bang that permeates the Universe

The light from these galaxies reached us after traveling for 13 billion years
Thus, we see 'young' galaxies formed within 1 billion years of "the Beginning"

Solar System Element Abundances



- H and He are by far most abundant elements
- Li, Be and B are anomalously low in abundance
- Exponential drop in abundance with increasing Z
- Even Z > odd Z
- Fe and neighbors are anomalously abundant

Nucleosynthesis in Stars



Al Cameron

PUBLICATIONS OF THE
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NUCLEAR REACTIONS IN STARS AND
NUCLEOGENESIS*

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INTRODUCTION

It was once thought that the stars and the interstellar matter had a uniform chemical composition except for some of the lighter elements, which were destroyed by thermonuclear reactions in stellar interiors. This view has caused astronomers and physicists to look for extreme physical conditions in which all the matter in the universe was gathered together at high density and raised to a high temperature sufficient to produce the observed abundances of the elements by the nuclear reactions that take place under these conditions. However, in recent years it has become apparent not only that thermonuclear reactions in stellar interiors can produce large abundance changes in even the heaviest elements,^{1,2} but also that there are intrinsic differences in the chemical compositions of different classes of stars before thermonuclear reactions have started in them.³ Stars classed as extreme Population II objects—subdwarfs and members of globular clusters—usually have a much smaller ratio of metals to hydrogen than does the sun, the factor of decrease being commonly about 10 or 20. In certain rare objects this factor may be much larger still.⁴ On the other hand, in the O- and B-type stars of extreme Population I the ratio of metals to hydrogen has commonly increased over that in the sun by factors of 2 to 4. Light and heavy elements

* One in a series of review articles currently appearing in the *Publications*.

"B²FH"

REVIEWS OF
MODERN PHYSICS

VOLUME 29, NUMBER 4

OCTOBER, 1957

Synthesis of the Elements in Stars*

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California Institute of Technology, Pasadena, California*

"It is the stars, The stars above us, govern our conditions";
(*King Lear*, Act IV, Scene 3)

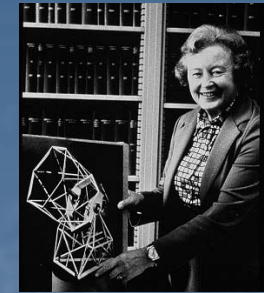
but perhaps

"The fault, dear Brutus, is not in our stars, But in ourselves,"
(*Julius Caesar*, Act I, Scene 2)

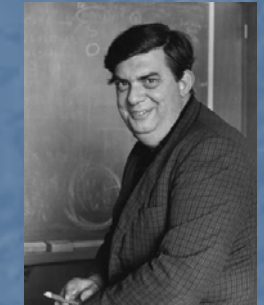
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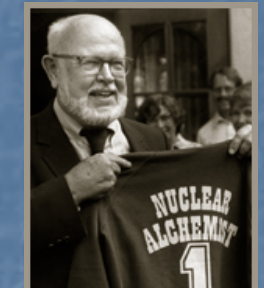
* Supported in part by the joint program of the Office of Naval Research and the U. S. Atomic Energy Commission.



Margaret Burbidge



Geoffrey Burbidge



William Fowler



Fred Hoyle

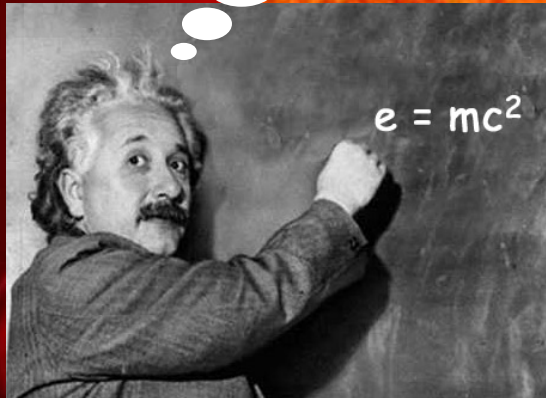
These two papers first explained nuclear reactions in stars responsible for the production of the elements


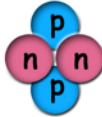
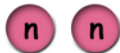
Energy Generation in Stars

The Sun's energy comes from the fusion of H into He

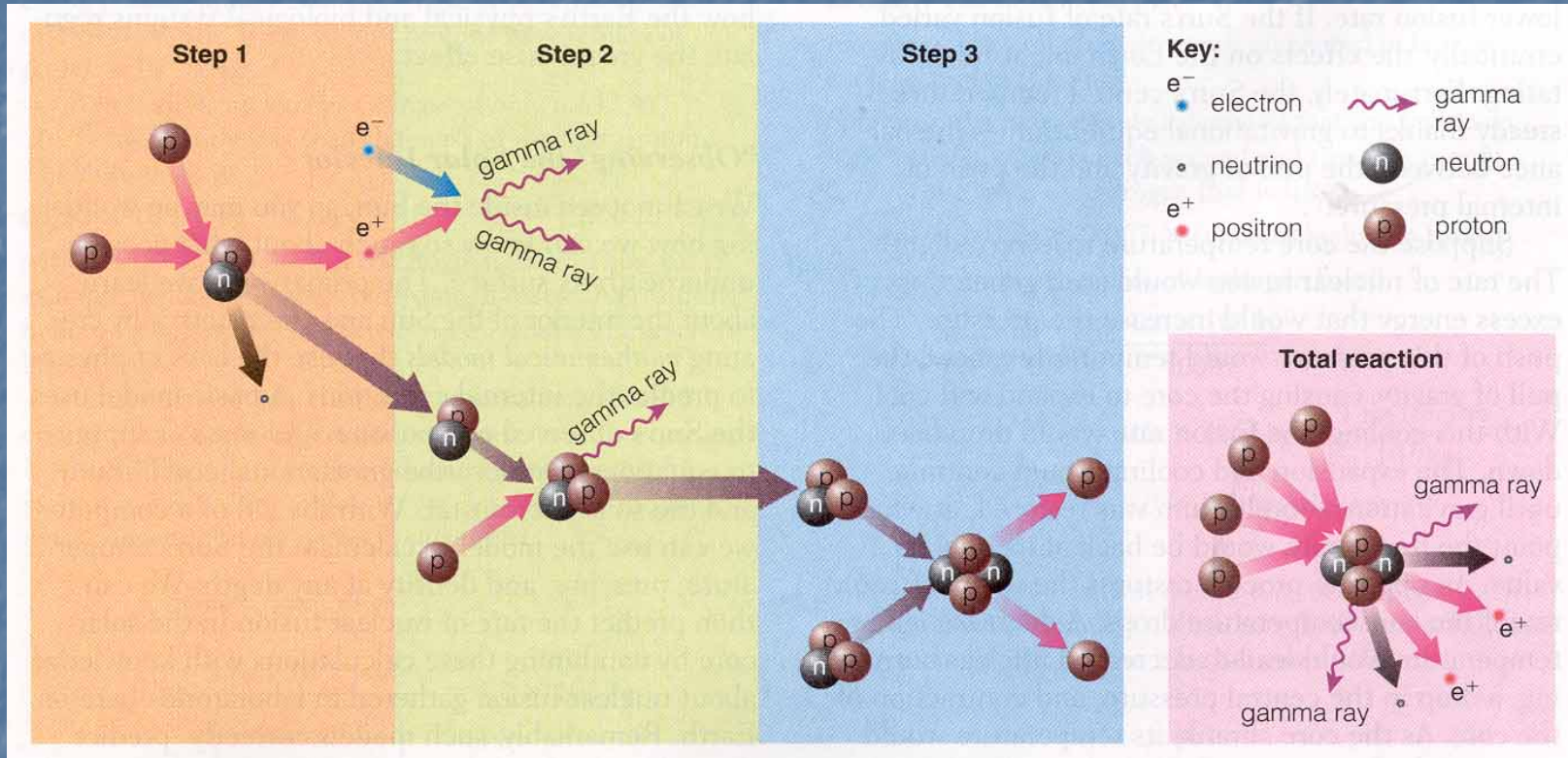
Energy can be created from mass!

The mass of a nucleus is less than the sum of the masses of the protons and neutrons that it is made from!



	mass = $2 \times 1.00728 \text{ u}$	
	mass = $2 \times 1.00866 \text{ u}$	
mass of $2p + 2n = 4.03188 \text{ u}$		mass = 4.00153 u
$\Delta m = 0.03035 \text{ u} = 5 \times 10^{-26} \text{ g}$		
$E = \Delta mc^2 = 28.3 \text{ MeV}$ <u>nuclear binding energy</u>		

Fusion of H into He



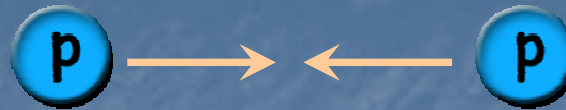
Reaction timescales

10^9 years

1 second

10^6 years

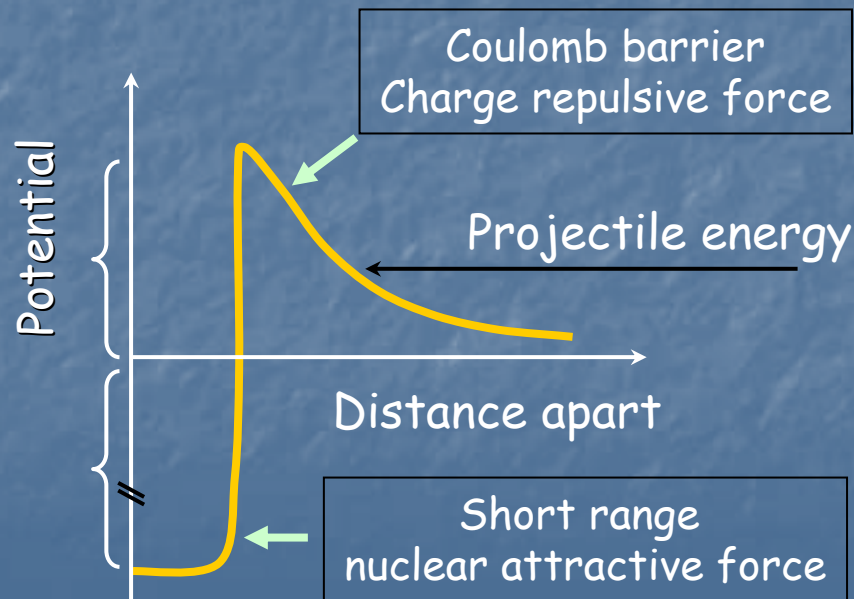
Nuclear Fusion



Nuclear binding force is strong but *very* short range

Nuclei must overcome Coulomb repulsive force to fuse together

Potential diagram for nuclear collisions



Temperature $\sim 15 \times 10^6$ K

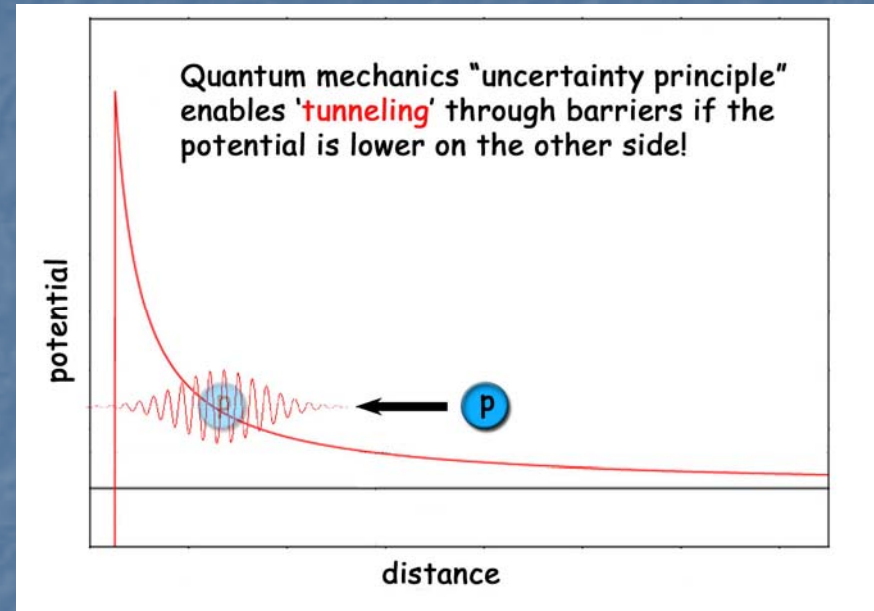
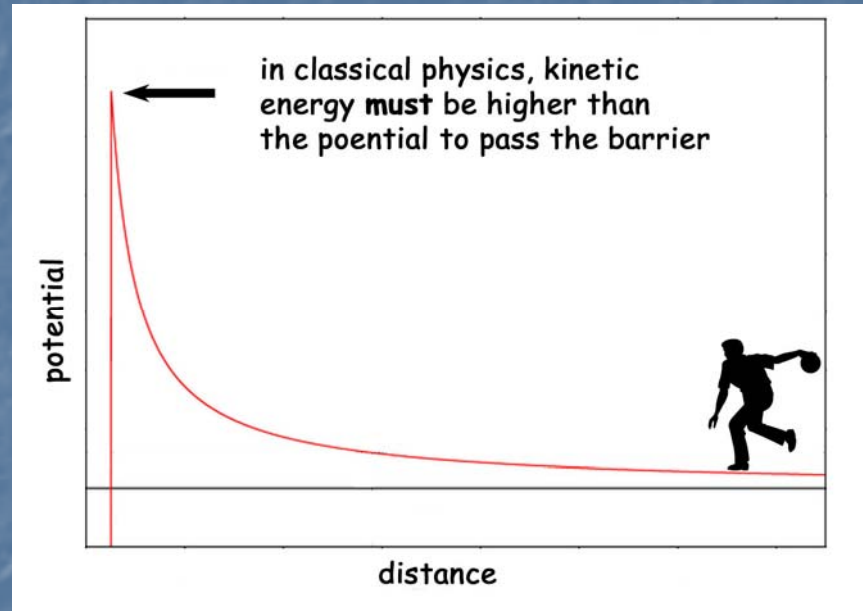
$$E = kT \sim 1 \text{ keV}$$

However!

p + p barrier $E = 550 \text{ keV}$

The sun is not hot enough for fusion ?!

Quantum Tunneling

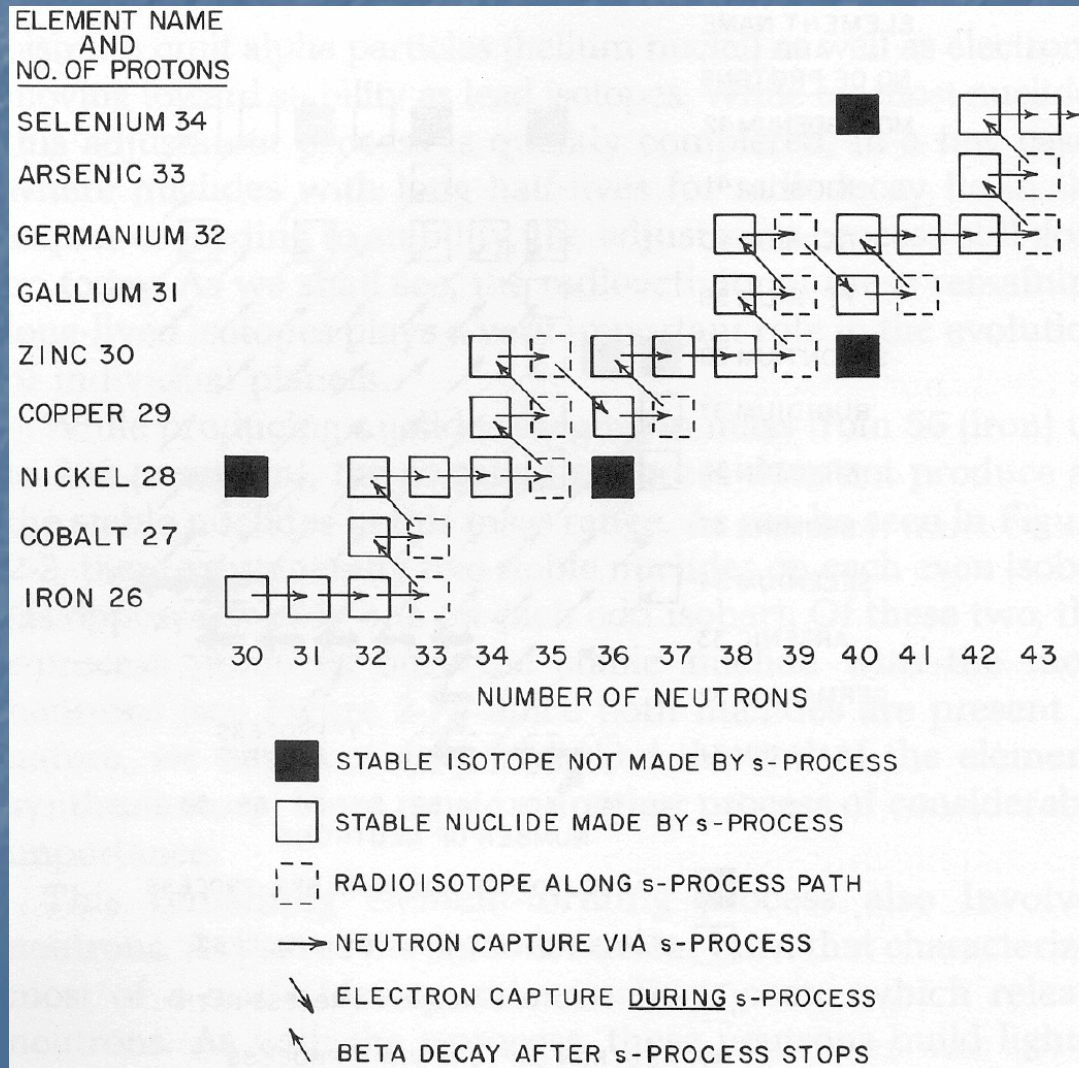


Probability of penetrating barrier $\sim \exp[-(E_{\text{barrier}}/E_{\text{thermal}})^{1/2}]$
Fusion reaction rates increase exponentially with Temperature

For $p + p$ reaction in the Sun, Probability = 10^{-22} !
This is why the Sun will live for 10 billion years

S Process Nucleosynthesis

slow neutron capture



Neutron capture reactions are *easy!*
 No charge barrier
 Can occur even at room temperature

Isotope with excess neutrons



S process is slow addition of neutrons compared with radioactive decay

S process is the most important source of many elements > Fe

This occurs at the late stages of a stars life

Core Temperatures in Stars

Stars are balanced in gravitational/thermal equilibrium

- Thermal pressure in core maintained by fusion reactions
- Evolution and lifetime of a star is “easy to predict”

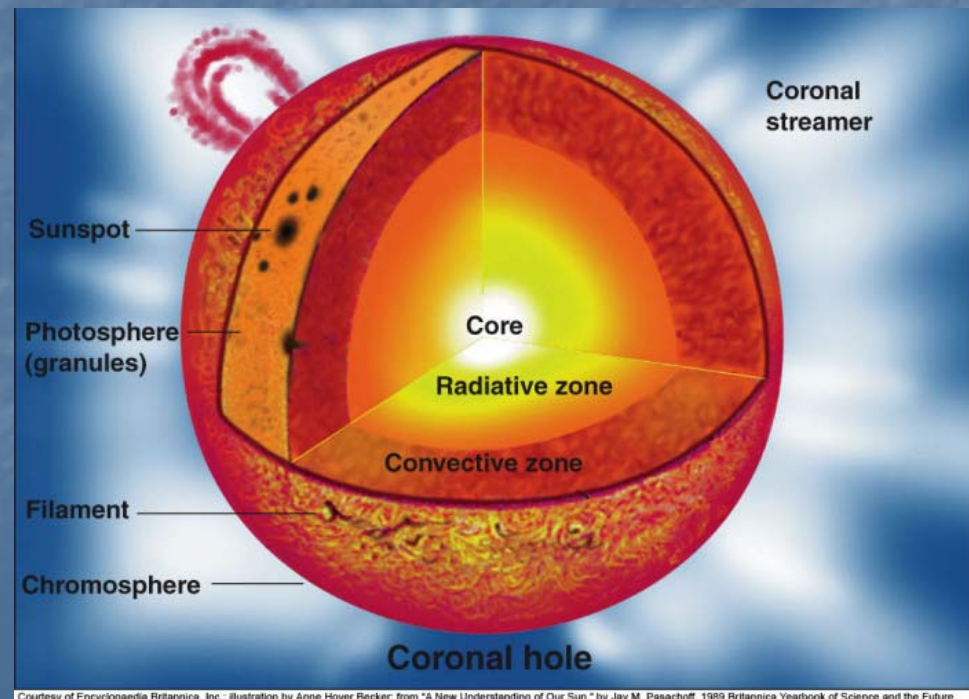
High mass stars have short lives

- Higher thermal pressure required to support higher mass
- Fusion rates increase exponentially with temperature

Pressure forces balance gravity force
(hydrostatic equilibrium)

- $F_{\text{gravity}} = P_{\text{gas}} + P_{\text{radiation}}$
- $mGM/R^2 = rkT + sT^4$

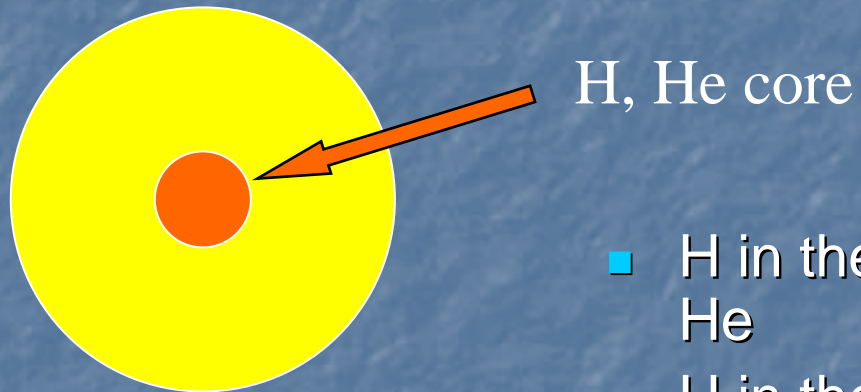
The temperature of the core is proportional to the mass of the star



Courtesy of Encyclopaedia Britannica, Inc.; illustration by Anne Hoyer Becker, from "A New Understanding of Our Sun," by Jay M. Pasachoff, 1989 Britannica Yearbook of Science and the Future

Stellar Evolution

Main sequence star

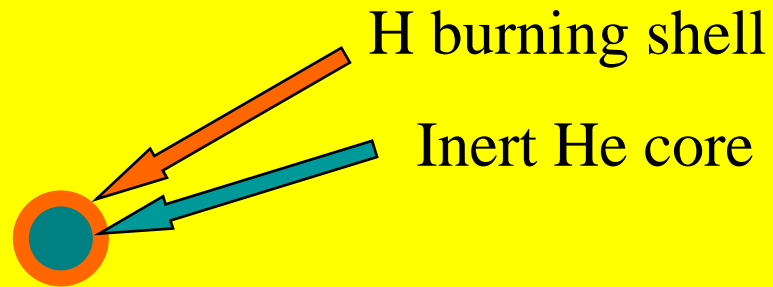


H, He core

- H in the core is gradually converted into He
- H in the core is eventually completely converted into He
- This is the longest lasting stage of a star's life
 - 10 million – 100 billion years, depending on stellar mass

Stellar Evolution

Red giant star



- Without H fusion, the He core cannot thermally support itself – the He core contracts
- Pressure, temperature increase in the core
- H burning starts on a shell around the core
- With higher core T, the star **EXPANDS**

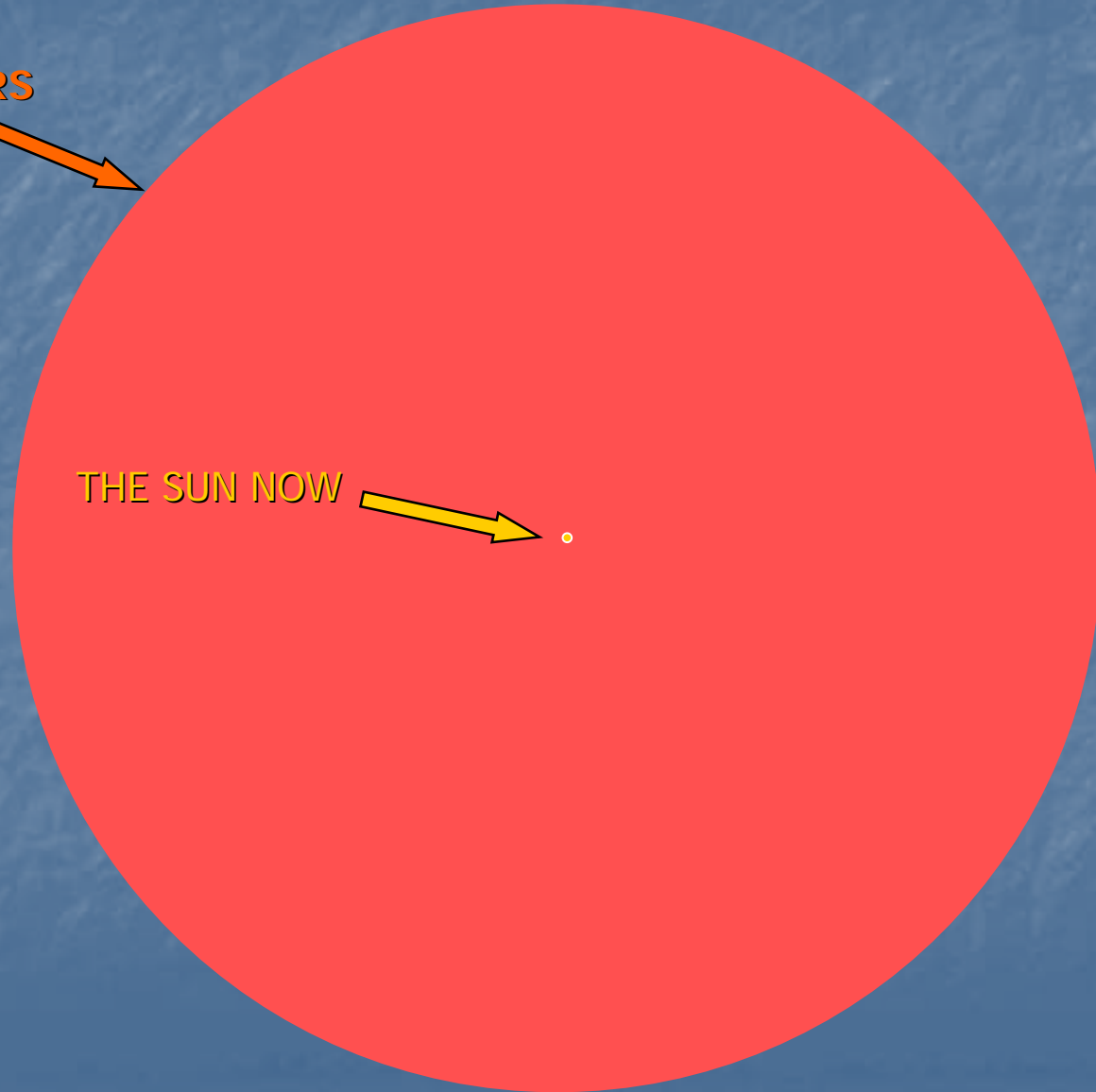
Red Giant Stars

THE SUN IN 5 BILLION YEARS

About the size of the Earth's orbit!

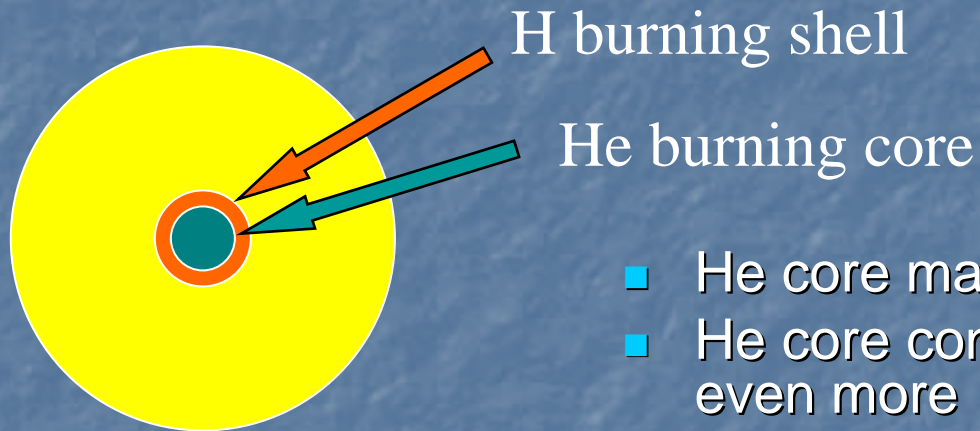
Diameter ~ 300 million km
Core diameter ~ 10,000 km
Core temperature 100 million K
Star luminosity 1,000 x the Sun

THE SUN NOW

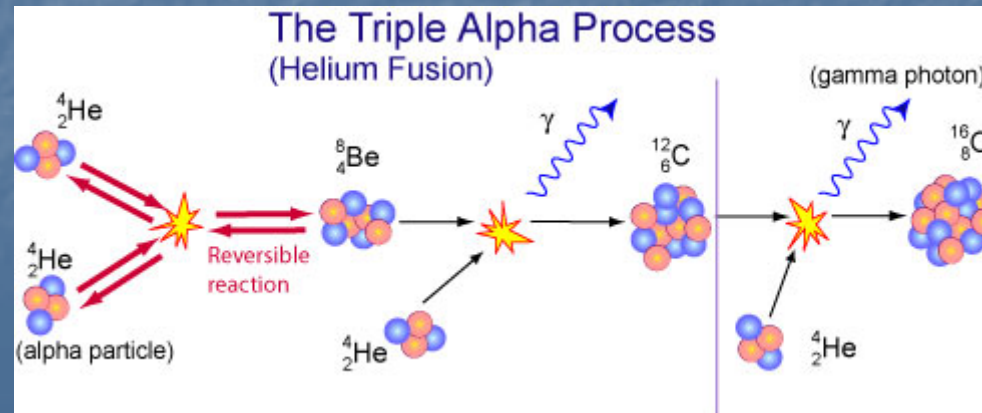


Stellar Evolution

Red giant star

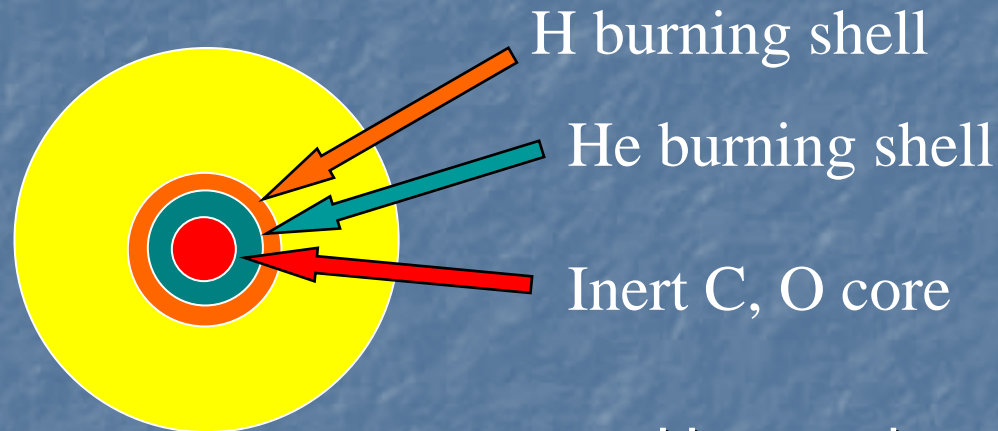


- He core mass continues to increase
- He core continues to contract, raising T even more
- Eventually T is high enough for Helium fusion reactions



Stellar Evolution

AGB star transition



(not to scale)

- He core is converted from He into C,O
- The star is now an “AGB star”
- He burns in a shell around the core
- S process elements made in He shell
- H burns in a shell around the He shell

Primary neutron source

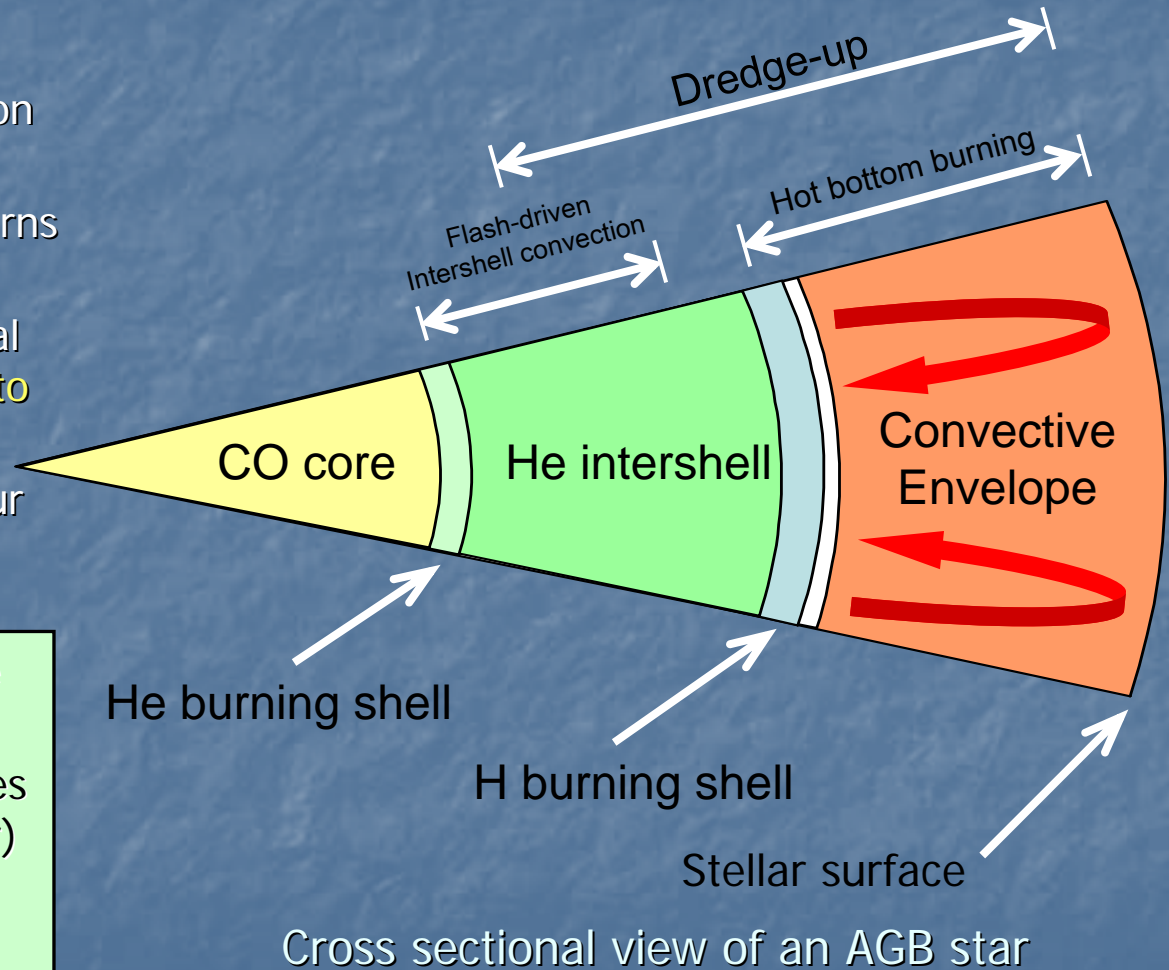


Stellar Evolution

AGB stars

- He shell gains mass from H fusion shell above
- He shell contracts and briefly burns strongly (flash)
- He flash creates a strong thermal pulse, **ejecting some material into space**
- He shell thermal pulses can occur many times

- He shell mixes with convective envelope
- C made from He burning makes the surface C-rich (carbon star)
- Thermal pulses eject stellar material into space
- The entire envelope is ejected, making a planetary nebula



Planetary Nebula

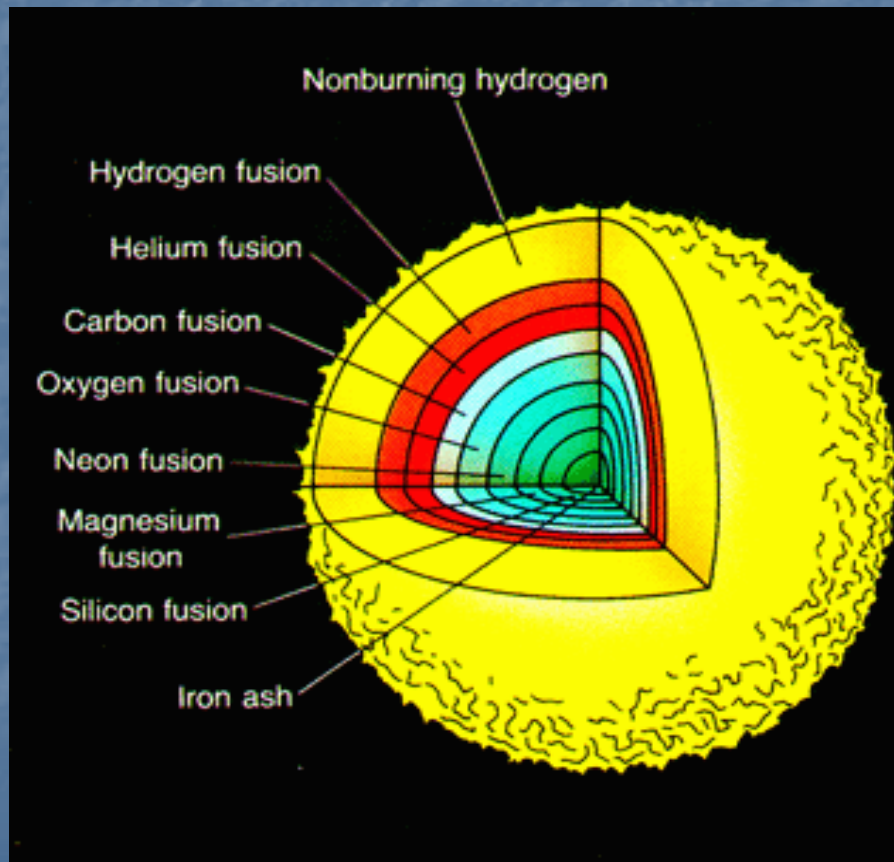
The fate of the Sun

Luminous gas and dust cloud expelled from AGB star
Central remnant is a white dwarf

Helix planetary nebula, Hubble telescope

Stellar Evolution

Massive Stars



High mass stars have core T high enough to ignite C,O fusion

Fusion of heavier elements continues at increasingly higher T, reaching 3×10^9 K for Si fusion

Fusion ends with Fe, which has the highest binding energy

Supernovae

Stand back!

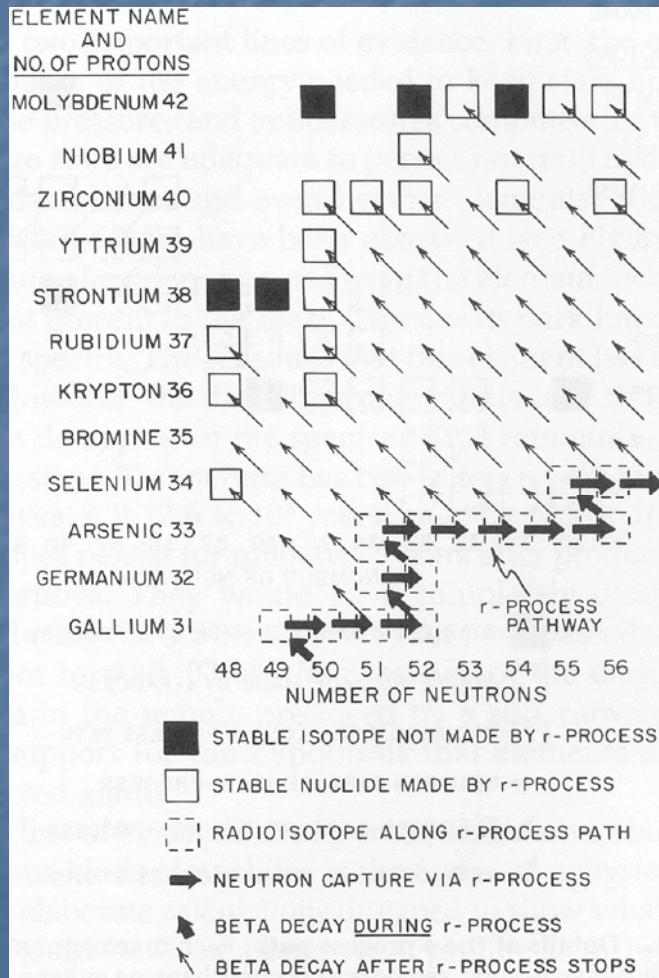
Fusion beyond Fe sucks energy from the star

Star collapses and then explodes

Explosion 100 billion times brighter than the sun

Explosive Nucleosynthesis

R Process



10^{12} times greater neutron density compared with s process


Very short-lived isotopes may capture a neutron before decaying

s process, r process have different products

Nucleosynthesis Summary

- All H and most He formed in the Big Bang
- He and all other elements form in stars
- Fusion reactions produce most elements up to Fe
- Heavy elements mostly made by s process and r process

The End is Also the Beginning



- Stars form from the remnants of previous generations of many other stars
- Most stardust resides in the Galaxy for a long time (10^8 years)
- In large star forming regions such as Orion (shown), massive stars may evolve and die (supernovae) while others are still forming



Solar Nebula

The starting materials

The Solar System was built from stardust

Almost everything has since been destroyed and mixed together

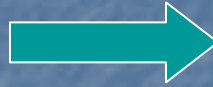
The Solar System now has a very uniform isotopic composition

Presolar grains (stardust) can be identified by their unusual isotopic ratios

Stardust in Meteorites



Murchison meteorite: well preserved sample of early Solar System



Noble gas carriers found by dissolving 99.9% of the meteorite in acid.

Noble gas carriers survived!

Discovery team:
Ed Anders, Ernst Zinner,
Roy Lewis, Sachiko Amari.



Noble gas with exotic isotopic compositions found in meteorites. Finding the carriers was tricky.

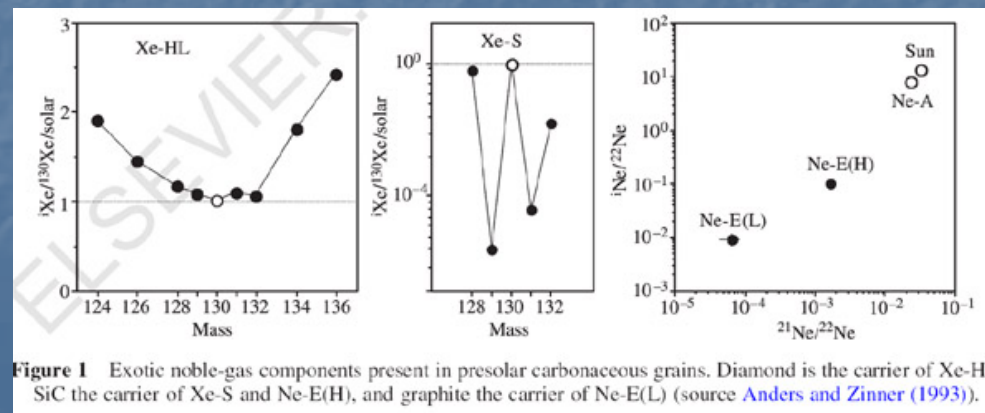
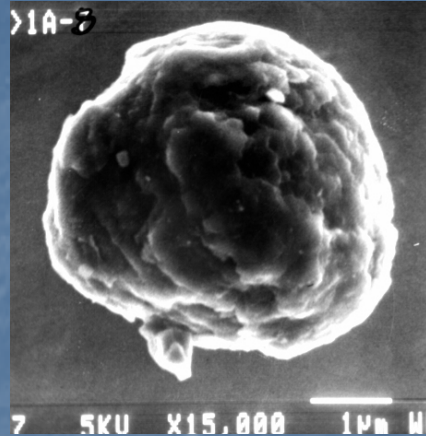


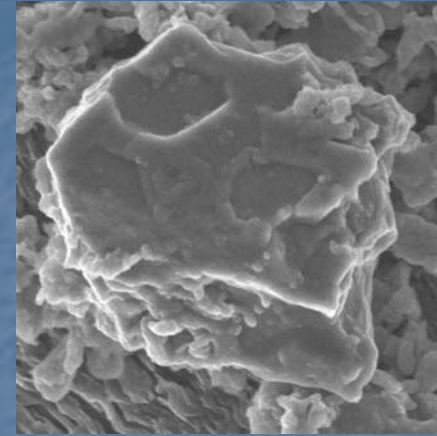
Figure 1 Exotic noble-gas components present in presolar carbonaceous grains. Diamond is the carrier of Xe-HL, SiC the carrier of Xe-S and Ne-E(H), and graphite the carrier of Ne-E(L) (source Anders and Zinner (1993)).

Stardust Extracted from Meteorites

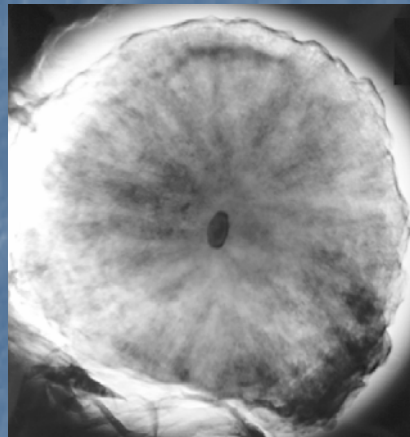
- SiC
- Si_3N_4
- Graphite
- Nanodiamonds
- Al_2O_3
- Hibonite
- Spinel
- Olivine
- Amorphous silicate



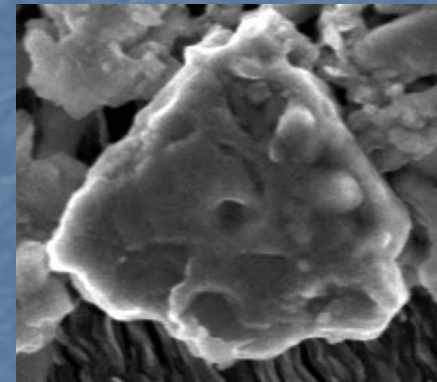
Graphite



Silicon Carbide



Carbides *Within* Graphite



Silicon Carbide

Typical size: 1 μm

Abundance: 1 – 100 ppm

Dust condensation in O-rich stellar outflows

T	Phase
1633	Corundum
1562	Hibonite
1537	Perovskite
1472	Melilite
1351	Spinel
1320	Plagioclase
1305	Forsterite
1287	Metal (Fe+Ni,Co,Cr,Si)
1250	Titanium Oxide
1246	Enstatite
1200	Cr-spinel (MgAl_2O_4 - MgCr_2O_4)
<1000	FeO

- Equilibrium thermodynamics (solar gas, 10^{-5} bars, after Yoneda & Grossman 1995; Ebel MESS-2)
- Identified as presolar grains
- No silicate stardust “rocks” found yet

Figure courtesy L. Nittler

Nittler, Alexander, Stadermann, Zinner (2005) MPS 40, 5208
Nittler, Alexander, Stadermann, Zinner (2005) LPS 36, 2200
Choi, Wasserburg, Huss (1999) ApJ 522, L133

Isotopic Signatures of Stardust

Major and minor element isotopic ratios vary by a factor of 10,000

Isotopic compositions reflect distinct nucleosynthetic processes

Identify stellar source, age, metallicity

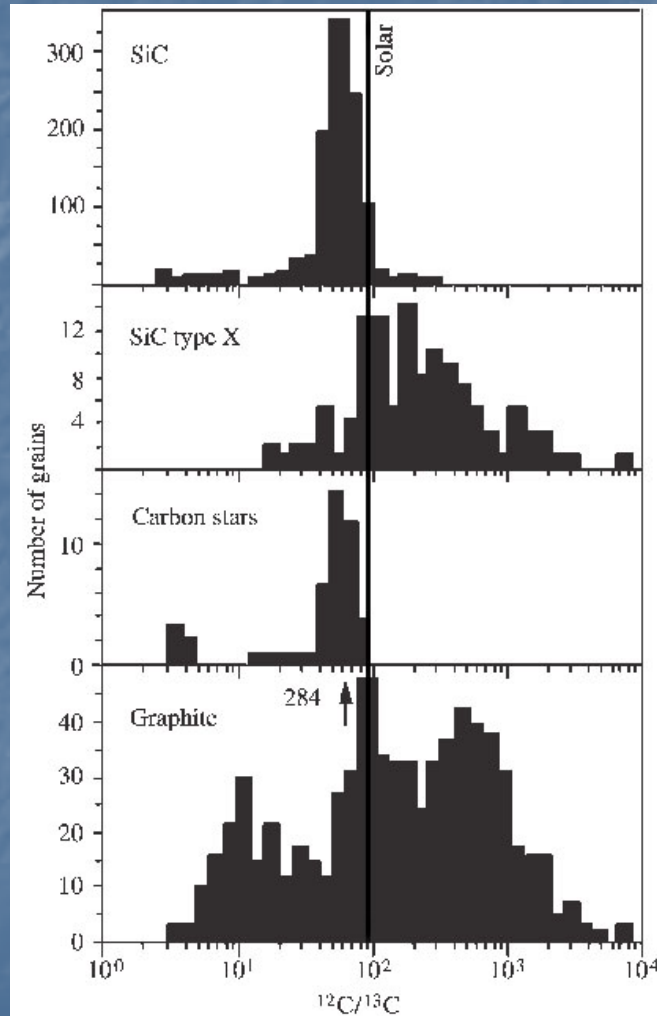


Figure courtesy E. Zinner

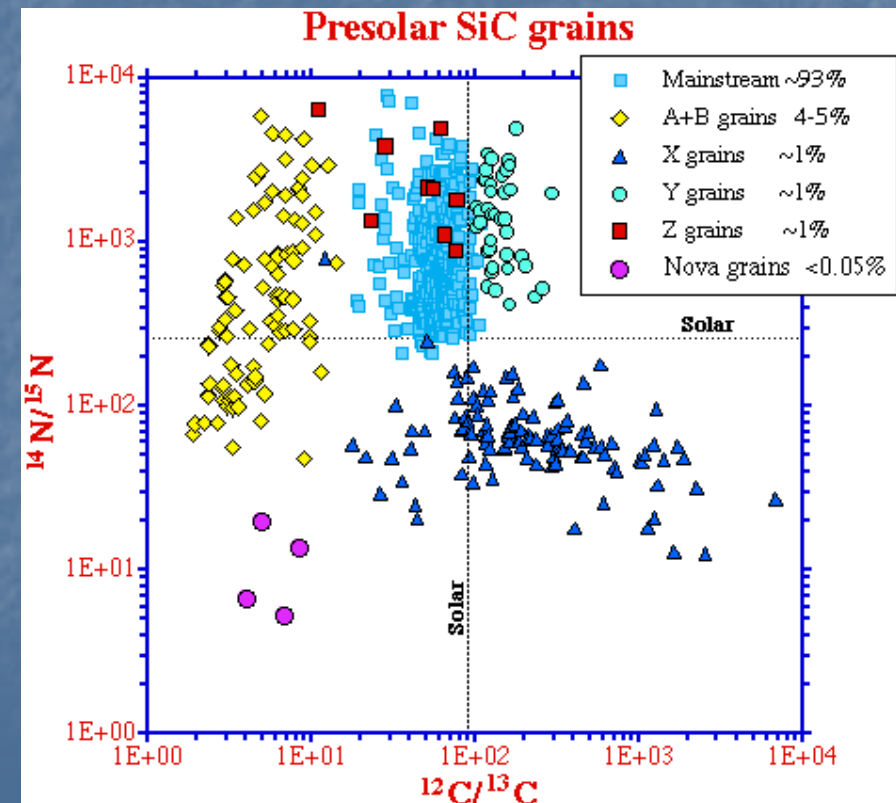


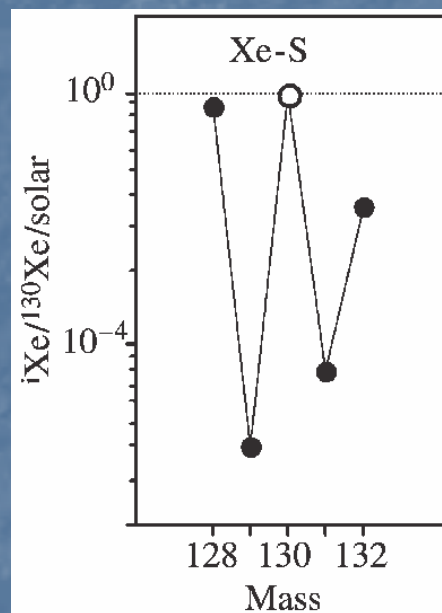
Figure courtesy E. Zinner

S-Process Signatures in SiC

Product of AGB stars

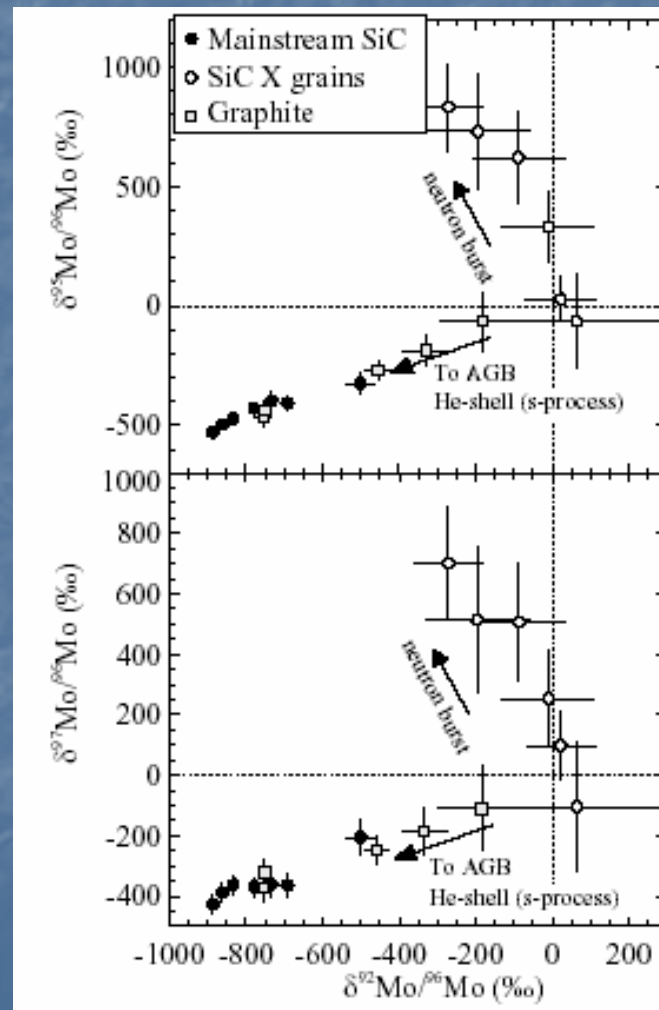
Bulk measurement
millions of grains

Anomalous Xe found in meteorite acid residues led to the chemical isolation of SiC and nanodiamonds



Anders & Zinner (1993) Meteoritics 28, 490

Single grain measurements



Savina et al. (2002) GCA

Short-Lived Nuclides from Supernovae

- ^{44}Ca and ^{49}Ti enrichments observed in rare SiC grains
- Grain must have formed shortly after nucleosynthetic production
 - *Requires stellar source*
- Resist loss or isotopic exchange since then (>4.5 billion years)
- *Identify extinct nuclides by isotopic anomaly in daughter product*

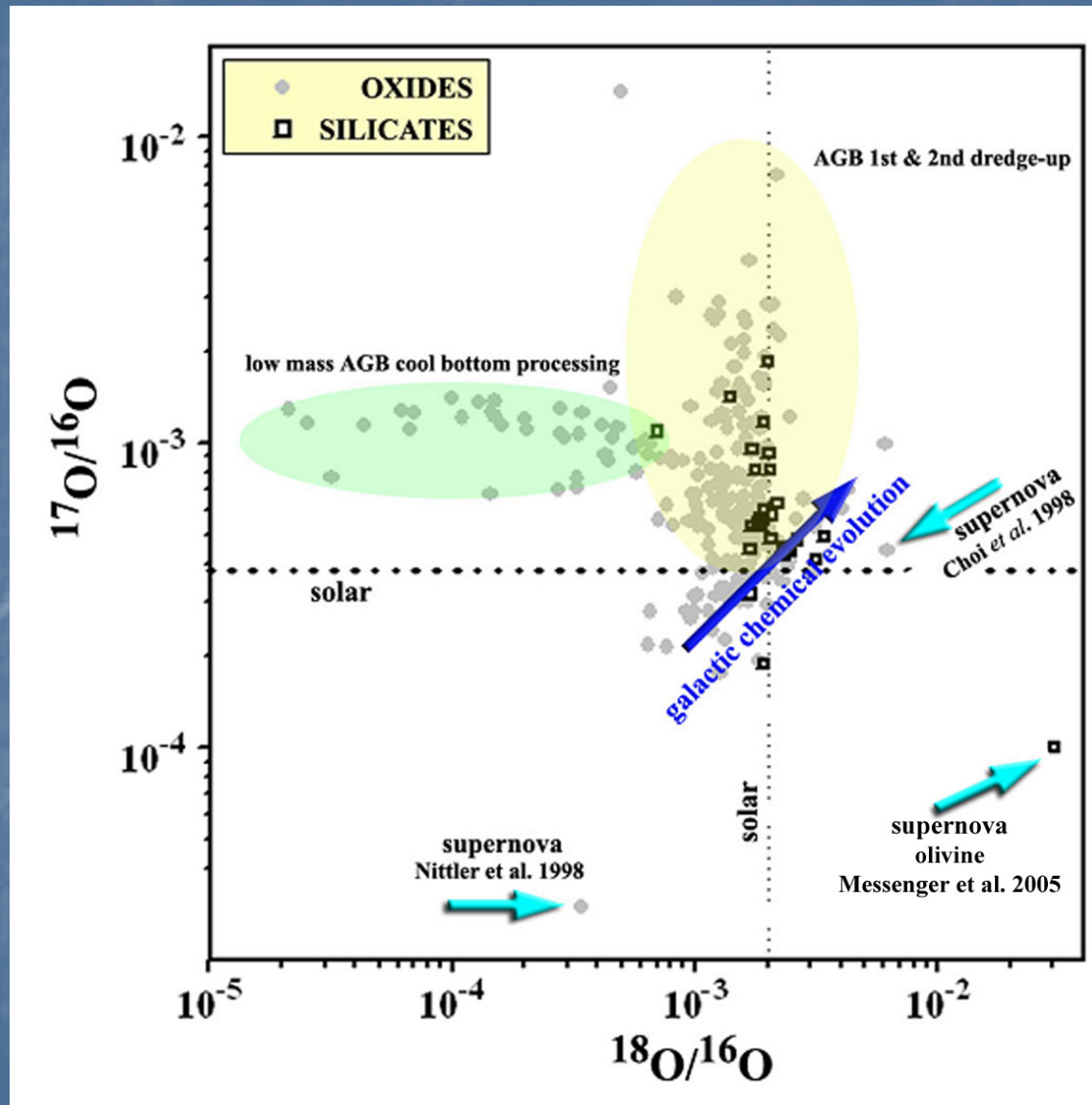
Example: ^{44}Ti

- produced only in supernovae
- $t_{1/2} = 60$ years
- $^{44}\text{Ti} \longrightarrow ^{44}\text{Ca}$
- $^{44}\text{Ca}/^{42}\text{Ca}$ ratios reach 5,000 x solar in stardust

Example: ^{49}V

- $t_{1/2} = 331$ days
- $^{49}\text{V} \longrightarrow ^{49}\text{Ti}$
- *Grains condensed within a few years of nucleosynthesis*

Silicate and Oxide Stardust



- >100 presolar silicates and oxides found so far
- ~99% originate from RG/AGB
- Supernova dust is rare (~1%)
- Galactic chemical evolution
- Estimate for age of the Galaxy

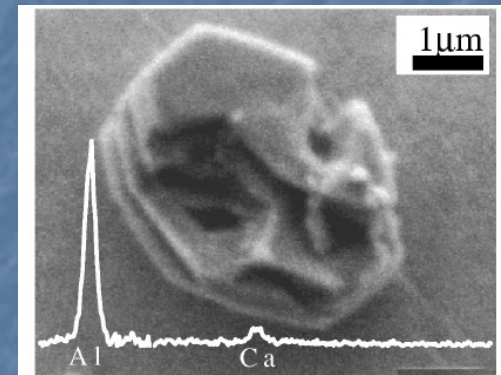


FIG. 1.—Secondary electron image of circumstellar hibonite S-H5323. The X-ray spectrum is shown by the white line. Note the platy structure and the crystal faces on each plate.

Floss & Stadermann (2005)
 Huss et al. (1993)
 Messenger et al. (2003) *Science*
 Nguyen & Zinner (2004) *Science*
 Mostefaoui & Hoppe (2004) *ApJ*
 Nagashima et al. (2004) *Nature*
 Nittler & Cowsik (2004) *Phys Rev*

Supernova Dust Formation

Supernova olivine grain

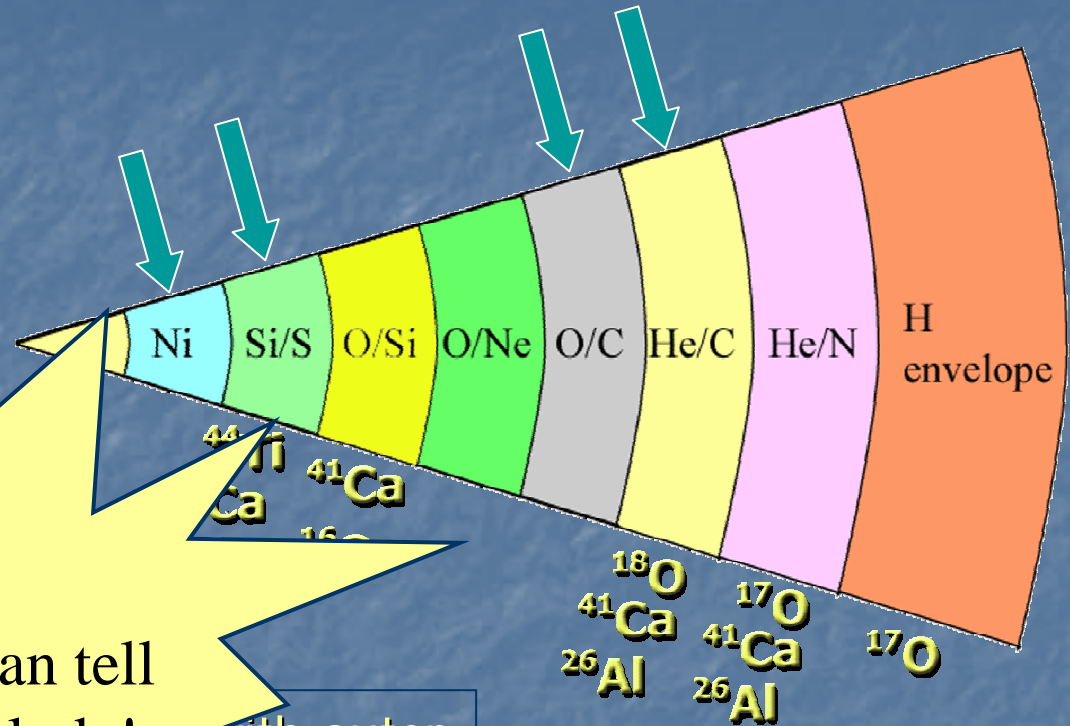
Olivine: $(\text{Mg,Fe})_2\text{SiO}_4$

$^{18}\text{O}/^{16}\text{O} = 13 \times \text{solar}$

$^{17}\text{O}/^{16}\text{O} = 1/3 \times \text{solar}$

$^{29}\text{Si}/^{28}\text{Si} = 0.8 \times \text{solar}$

^{18}O -enrichment



A 500 nm grain can tell us how stars explode!

Model re-
layers v.

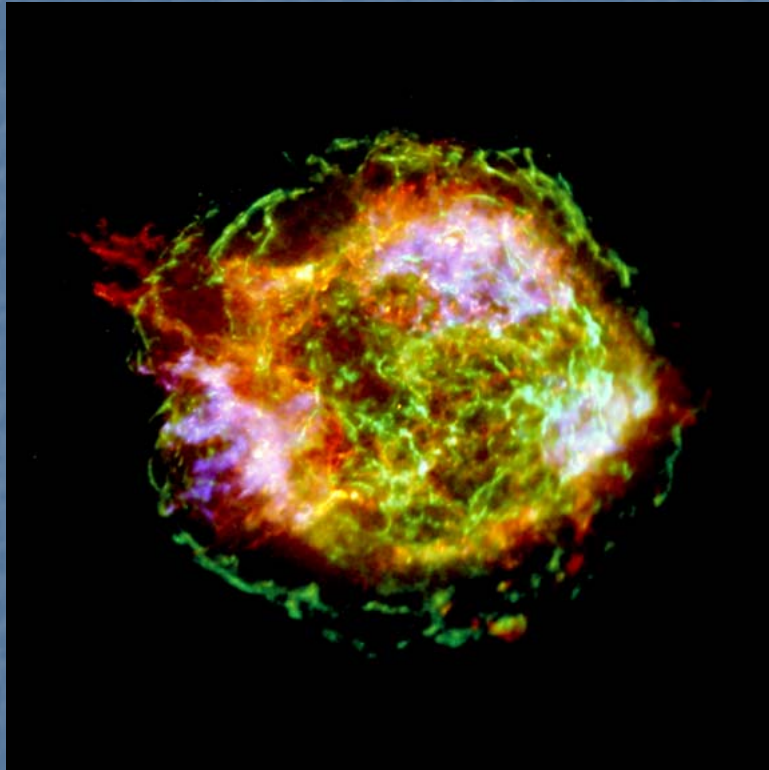
with outer
ing zones

Figure courtesy L. Nittler

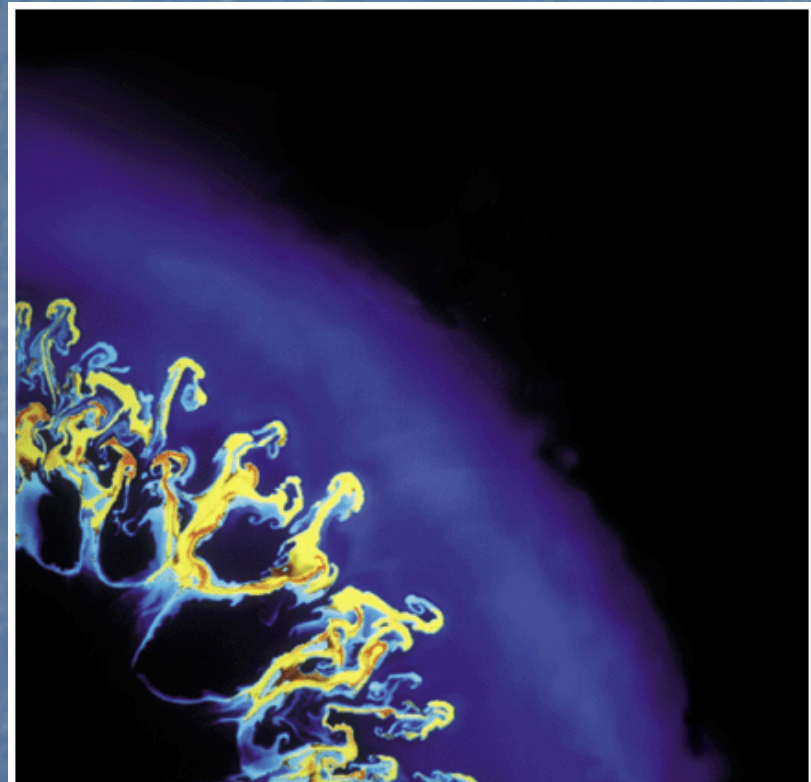
Messenger et al. (2005) Science

See also supernova SiC & graphite: Nittler *et al.* 1996, Hoppe *et al.* 2000

Theoretical and Observational Evidence for Complex Mixing in Supernovae



Chandra x-ray image of Cassiopeia A Supernova remnant



SN 1987A Muller et al. *A&A* 251, 505 (1991)

Stardust Summary

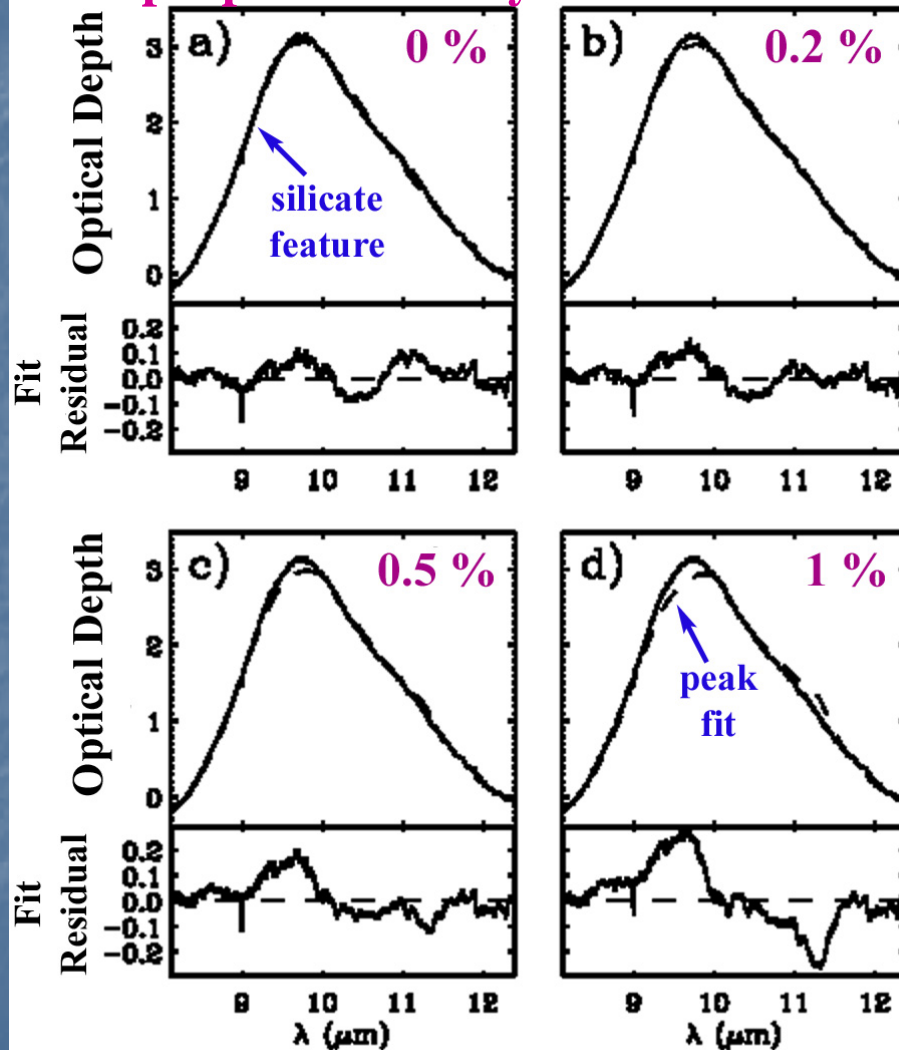
- Meteorites and cosmic dust contain ~100 ppm of μm -size stardust
- 99% originate from red giant/AGB stars
- 1 % originate from novae and supernovae
- Each of the nucleosynthesis processes are recorded among these stardust grains

Stardust Frontier: Silicates

- Silicates are abundant but only recently found
 - Huge background of Solar System silicates
- Commonly observed throughout the Galaxy by astronomers
- Silicates are sensitive recorders of their history
 - Radiation, thermal processing, aqueous alteration

Silicate Stardust Evolution

proportion of crystalline silicates



Kemper, Vriend, & Teilens (2004) ApJ 609, 826

Silicate mineralogy from 10 μm feature

Evolved stars: 10 – 20 % crystalline silicates

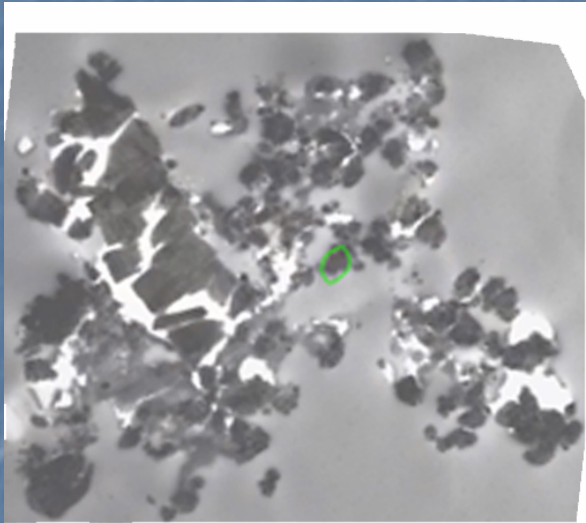
Diffuse ISM: ~ 1 % crystalline silicates

**Young stars: mix of crystalline/amorphous
& Comets & IDPs**

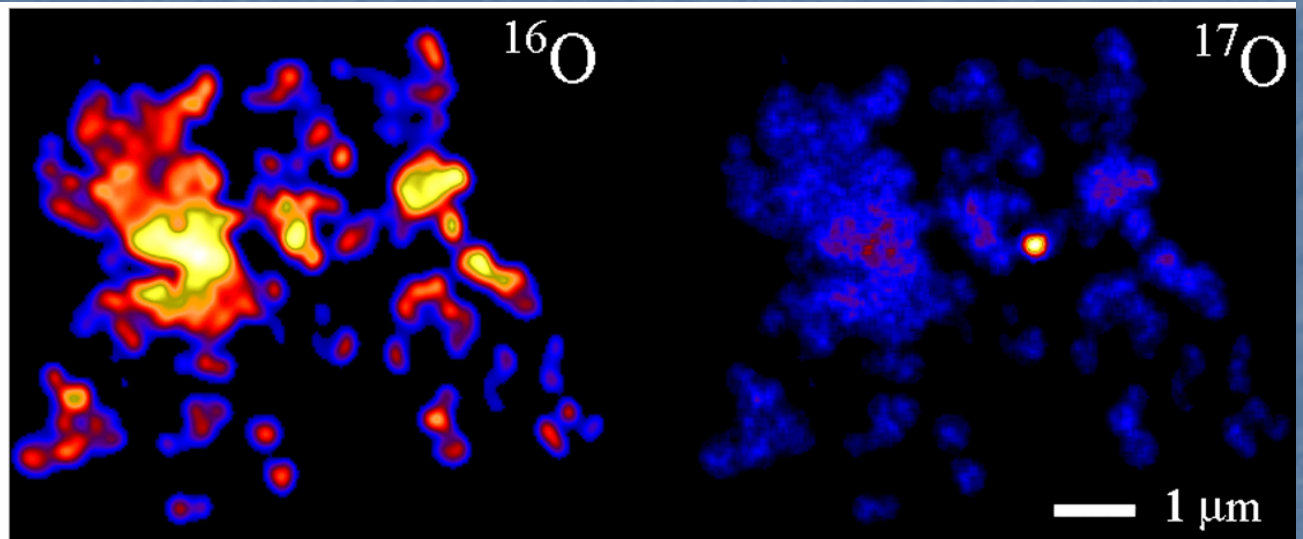
Meteorites: crystalline silicates dominate

Crystalline silicates are rendered amorphous or destroyed on ~ 10 Ma timescale. Re-formed near young stars

Found! Silicate Stardust in an Interplanetary Dust Particle



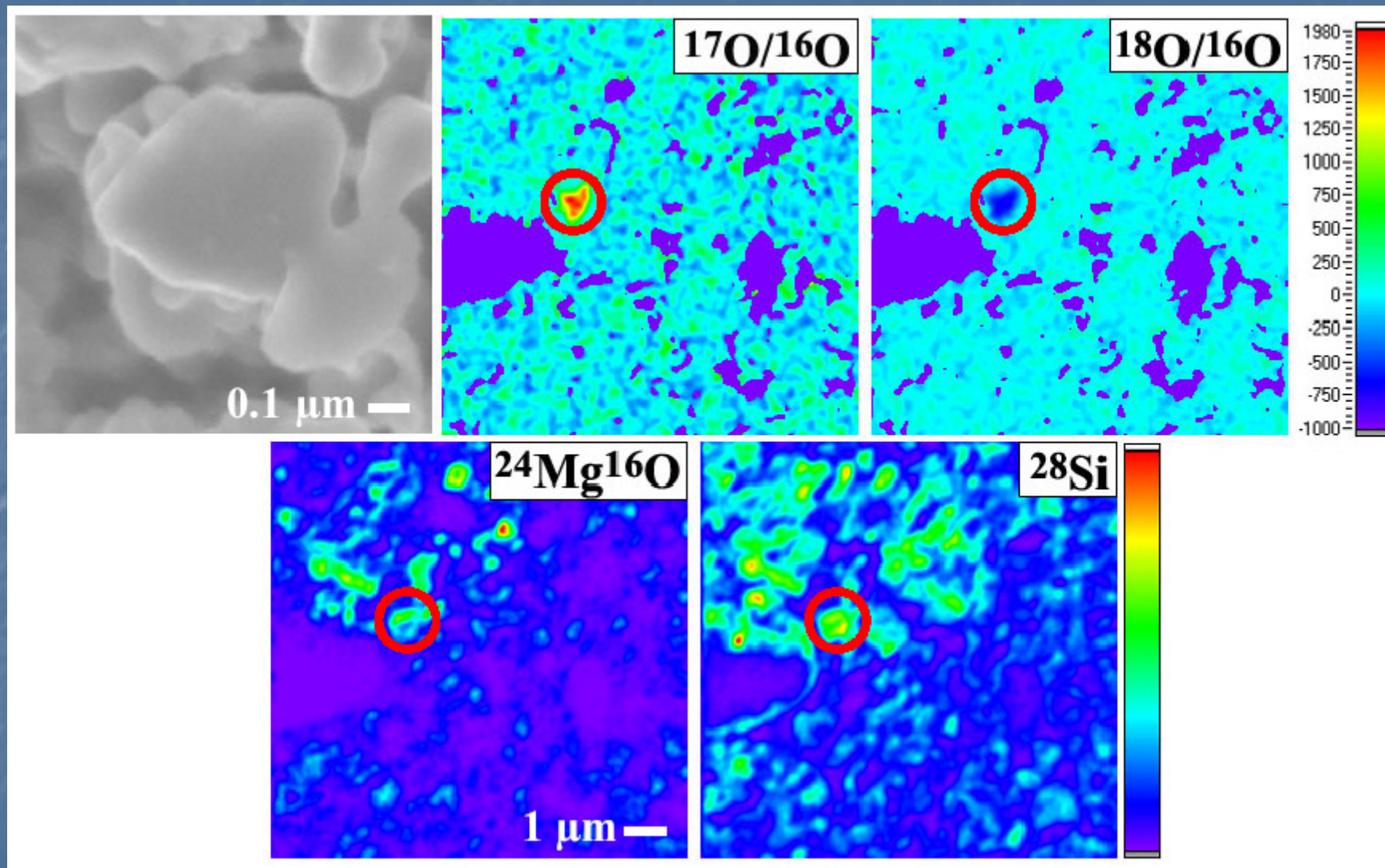
TEM Image



NanoSIMS ion images

Silicate stardust grains are small ($<0.5 \mu\text{m}$) and hidden within an overwhelming background of solar system silicates

Presolar Silicate in Acfer 094



Nguyen & Zinner, Science, 2004

Identifying Interstellar Dust in IDPs

GEMS

Enstatite

Forsterite

FeS

Ca,Al-rich glass

Thermally altered silicates

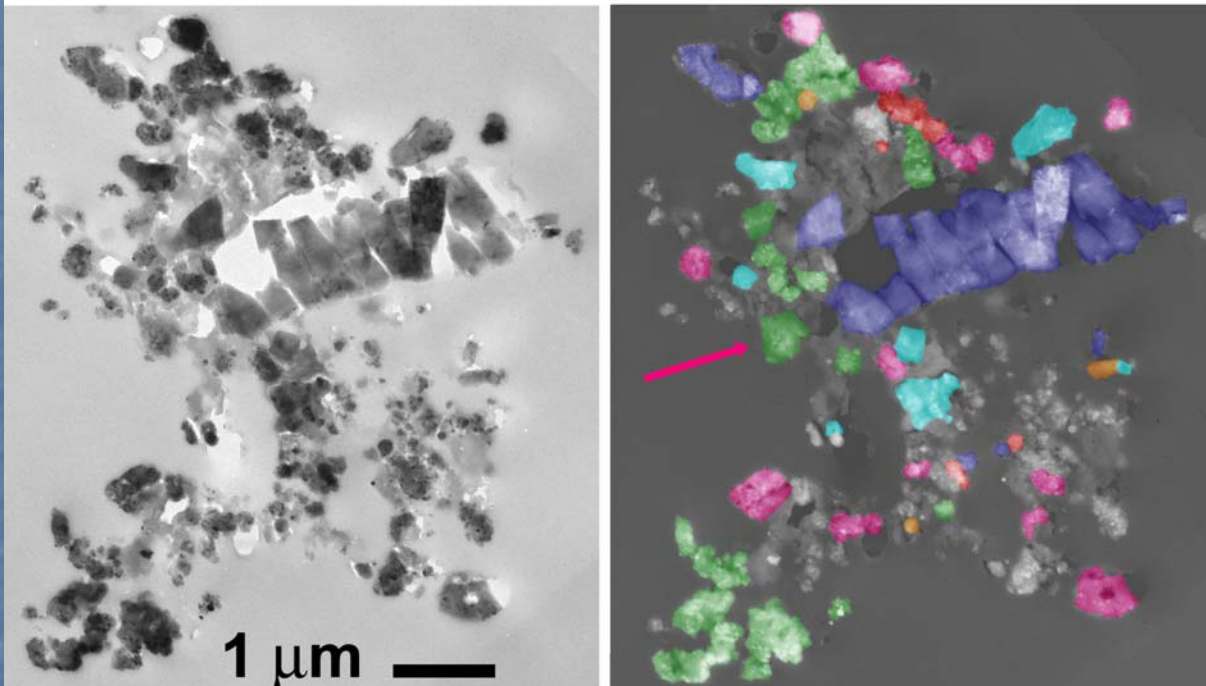
Presolar grain abundances:

4 of >100 GEMS

2 of >40 forsterite grains

0 of >40 enstatite grains

0 of 5 anorthite grains

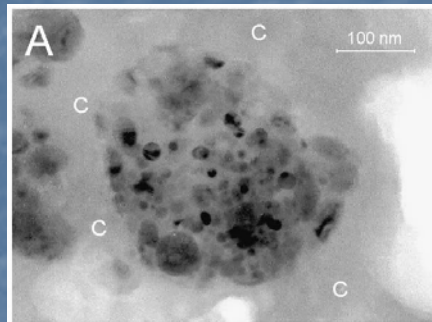


Thin section of IDP studied by TEM with typical anhydrous mineralogy

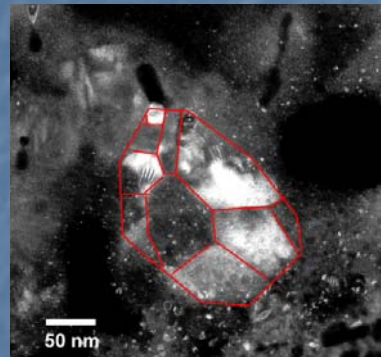
Presolar Silicate Mineralogy

- >150 silicate stardust grains

- 8 amorphous silicates



- 2 Olivine grains



amorphous:crystalline silicate abundances

- **Interstellar dust: 100:1**
- **in meteorites, IDPs: 4:1**

Most amorphous interstellar dust is missing

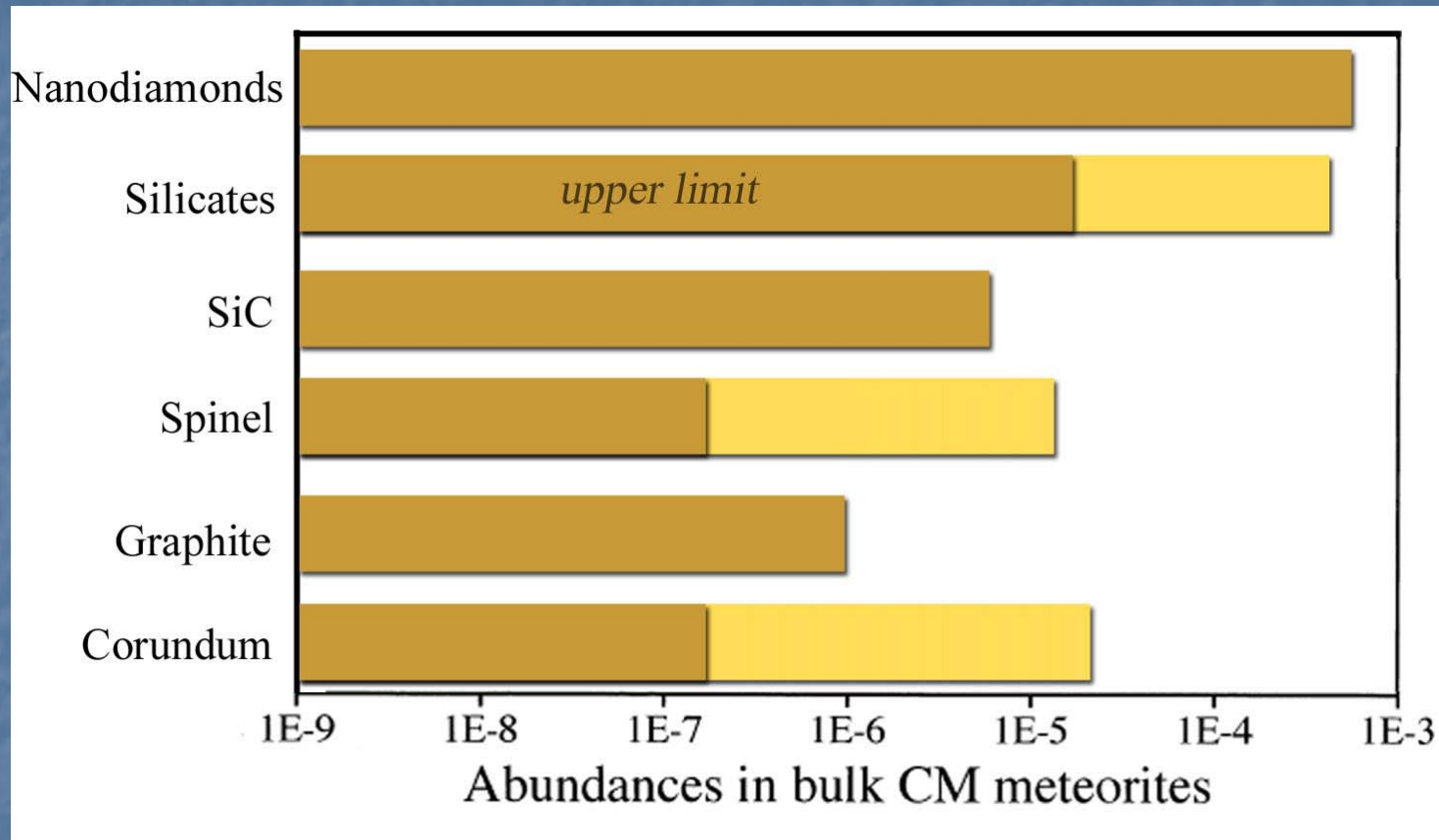
- Amorphous grains more easily destroyed?
- Annealed into crystalline silicates?
- Not isotopically distinct?

Nguyen et al. (2007) *ApJ*, *in press*

Messenger et al. (2003) *Science* 300, 105

Nguyen & Zinner (2004) *Science* 303, 1496

Stardust Abundances



New ability to probe smaller grains (200 nm)

O-rich stardust grains now known to 'outweigh' carbonaceous phases

Nguyen et al. (2007) ApJ, *in press*

Zinner (2003) GCA 67, 5083

Nittler (2003) EPSL 209, 259

Messenger et al. (2003) Science 300, 105

Stardust Frontiers

- Ancient materials preserved in meteorites are direct samples of evolved stars and supernovae
- >10 to 100 stars seeded our own solar system
- New samples of cometary material now being searched for stardust
- Have comets better preserved the Solar System starting materials?



First look at the Stardust samples